

An Introduction To The Study Of Wave Mechanics By Louis De Broglie

By Louis De Broglie

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Introduction. In 1923, Louis de Broglie, Since de Broglie believed particles and wave have the same Through the equation (λ) , de Broglie substituted \

http://chemwiki.ucdavis.edu/Physical_Chemistry/Quantum_Mechanics/02_Fundamental_Concepts_of_Quantum_Mechanics/De_Broglie_Wavelength

6. Particle- Wave Duality | Structure of the Atom -

and resources for further study. 6. Particle-Wave Duality Heisenberg uncertainty principle, Schrödinger equation: Keywords: Louis de Broglie,

<http://ocw.mit.edu/courses/materials-science-and-engineering/3-091sc-introduction-to-solid-state-chemistry-fall-2010/structure-of-the-atom/6-particle-wave-duality/>

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De Broglie L. An Introduction to the Study of -

Methuen & Co. Ltd., first published in 1930. - 253 pp. The lecture course given by Louis de Broglie, one of the founders of the quantum mechanics.

<http://www.twirpx.com/file/1034009/>

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Quantum mechanics is the science of the very small: 8 Dirac wave equation; 9 Quantum entanglement; de Broglie, Louis

http://en.wikipedia.org/wiki/Introduction_to_quantum_mechanics

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<http://philpapers.org/rec/DEBAIT>

The-History-of-the-Atom - Louis de Broglie -

Louis de Broglie was born in Dieppe, France on 15th August, 1892 and died on March 19, His ideas were a basis for developing the wave mechanics theory.

<http://the-history-of-the-atom.wikispaces.com/Louis+de+Broglie>

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